COMPONENTS: ORIGINAL MEASUREMENTS: (1) 1-Octyne; C_8H_{14} ; [629-05-0] McAuliffe, C. J. Phys. Chem. 1966, 70, 1267-75. (2) Water; H₂O; [7732-18-5] VARIABLES: PREPARED BY:

EXPERIMENTAL VALUES:

One temperature: 25°C

The solubility of 1-octyne in water at 25°C was reported to be 24 $q(1)/10^6$ q(2).

The corresponding mass percent and mole fraction, $\boldsymbol{x_1}$, calculated by the compilers are 0.0024 g(1)/100 g sln and 4.4 \times 10⁻⁶.

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

In a 250-mL bottle, 10-20 mL of (1) was vigorously shaken for 1 hr, or magnetically stirred for 1 day, with 200 mL of (2) at 25°C. The bottle was set aside for 2 days to allow droplets of undissolved (1) to separate. Absence of emulsion was checked microscopically. A sample of the hydrocarbon-saturated water was withdrawn with a Hamilton syringe and gas liquid chromatographed in conjunction with a flame- ESTIMATED ERROR: ionization detector.

SOURCE AND PURITY OF MATERIALS:

(1) Phillips Petroleum or Columbia Chemical; used as received.

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(2) distilled.

temp. ± 1.5°C soly. $0.8 g(1)/10^6 g(2)$ (standard deviation of mean)

REFERENCES: